## Even More Practice on Structures & Formulas!

KEY

- 1. Redraw in the style as indicated
  - (a) Draw in expanded style: HO<sub>2</sub>CCO<sub>2</sub>H

H-0-2-6-0-H

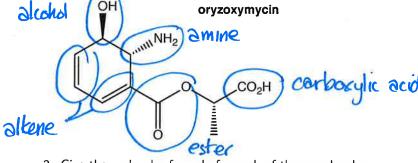
 $\sqrt{N}$ 

L 2 carboxylic acids connected together

 $(CH_3CH_2)_2N(CH_2)_2CH_3$ 

- (b) Draw as a condensed formula:
- 2. Circle and name the functional groups present in this molecule.

or  $CH_3(CH_2)_3N(CH_2CH_3)_3$ 



- 3. Give the molecular formula for each of these molecules.
  - (a) CH<sub>3</sub>O(CH<sub>2</sub>)<sub>2</sub>O(CH<sub>2</sub>)<sub>2</sub>OCH<sub>3</sub>

C6 41403



C5H5N

- 4. Calculate the IHD and draw three isomers for each formula.
  - (a)  $C_5H_{12}O$

 $1 + D = 5 - \frac{1}{2}(12) + 1 = 0$ 

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(b)  $C_2H_3N$ 

$$1HD = 2 - \frac{1}{2}(3) + \frac{1}{2}(1) + 1 = 2$$

(a)

CH3-CEN



Hzc=C=NH

5. For each pair below, state whether the two representations are the same molecule, or different molecules.

same (flip around both horizontal & vertical axes)

different (formula is different)